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Solutions and Mixtures Lab 1. What is a solution? 2. What is a mixture? Lab: 1) Using seven beakers of room temperature water, place the following item in a separate beaker. 2) Watch closely and determine if each is a solution or a mixture. 3) Explain why the two substances produce a solution or mixture.

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### **Mixtures And Solutions Interactive**

mixtures. Atoms are the basic building blocks for all three types of matter. Elements, compounds, and mixtures differ in the way that atoms are combined. ACTIVE READING 5 Compare What do elements, compounds, and mixtures have in common? ts asty ich, atoms are the ts” that . 6 Claims

- Evidence
- Reasoning If you have ever baked a cake

### **LESSON Pure Substances and Mixtures - Ivy Hawn**

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I then let the students know that specifically we are going to look at mixtures over the next few days. I explain that in the past lesson, Marshmallow Molecules, we looked at how mixtures are not combined chemically and can be separated by physical means. In addition, we read about how when solutes are dissolved in a solvent, a mixture is created.

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